

Sustainable Ingenuity: Exploring the Supremacy of Green Synthesis in Nanoparticle Fabrication Compared to other conventional Methods

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Abstract

Nanotechnology has surged in popularity all over the years. Nanoparticles are the essential component of nanotechnology. Nanoparticles are tiny particles that range in size from 1 to 100 nanometres and are made of carbon, metal, metal oxides, or organic substances. At the moment, nanoscale metals are mostly synthesized chemically, which has unexpected consequences such as excessive radiation and the use of extremely concentrated reductants and stabilizing agents that are hazardous to the environment and human health. In response to these obstacles, a comparatively new and largely unresolved area of study concentrating on synthesizing nanomaterial by biological method has arisen. Nanoparticles synthesized utilizing biological processes, often known as 'Green synthesis,' have a variety of properties, including higher stability and suitable diameters. Green synthesis employs ecologically beneficial, non-toxic, and risk-free chemicals. Hence, due to existing environmental risks and toxic wastes linked with chemical synthesis, green synthesis presents alternative advancement opportunities.

Keywords: nanoparticles, green synthesis, physical and chemical methods, advantages

Introduction

Nanotechnology is a field of science concerned with the analysis of materials in the nanoscale, typically ranging from 1 to 100 nm [1]. In terms of comparing to real items, an example that hair is 150,000 nanometers in diameter can be assumed. Nanotechnology is a rapidly emerging multidimensional research and advanced field of biology, chemistry, physics, food, medicine, electronics, aerospace, medicine etc., that evaluates the design, manufacture, assembly, and assessment of materials lesser than 100 nanometers in magnitude, along with the application of miniature functional structures made from these materials. It encompasses all development efforts.

Nanobiotechnology, in contrast, is the effect of an integration of biotechnology and nanotechnology fields with a shared functioning [2]. Various ways are used to create nanoparticles. There is an expanding amount of research focused on the manufacturing of nanoscale metals utilizing chemical, physical, and green synthesis methods [3,4]. Chemical and physical methods are progressively being replaced by green synthesis [5] due to the problems associated with huge quantities of energy consumed [4], the generation of potentially hazardous chemicals [6] and the use of complicated instruments and the production conditions [7,8]. Aerosols [9], ultraviolet (UV) rays [10] and thermal breakdown [11] are all physical processes that involve high temperatures as well as pressures [11].

The aerosol technique, for instance, involves a flame temperature of roughly 2400 K to create atomized aerosol crystals and nanoparticulate metals. The formation of PdO NPs via plasma-assisted physical vapor condensation necessitates three heat cycles around 250 °C and 800 °C, resulting in massive energy utilization. Chemical methods always involve sodium borohydride, a costly and hazardous reagent, alongside various dispersion stabilizers and organic solvents [12].

Green synthesis, on the other hand, uses natural and ecologically advantageous components (for example, reducing agents). Certain green elements may be utilized as both end-capping agents and dispersants [13] that not solely saves energy but effectively prevents the use of poisonous and dangerous chemicals. Currently, green synthesis is primarily based on microorganisms (fungi, bacteria, and algae) [14,15] or herbal extracts from leaves [13,16,17], flowers [18,19], roots, peels [20], fruits [21] and seeds [22,23] of many plants. Green products comprise polyphenols and proteins, which can be used instead of chemical reagents as reducing agents to decrease metallic ions to a decreased valence level [24]. Metal nanoparticles can be synthesized with the help of green materials within the appropriate circumstances (temperature,

concentration, ambient air). Under some situations, the effectiveness of green synthesized metal nanoparticles surpass that of chemically synthesized metal nanoparticles. For instance, the particle size of Fe₃O₄ NPs synthesized using the green synthesis approach is 2-80 nanometers, which is extensively lower than the 87-400 nm particle size of particles created using the wet chemical technique [25]. The creation of nanoparticles via physical and chemical processes has been reported in prior research.

This review outlined the techniques involve in the synthesis of nanomaterials, as well as the key concerns and issues in traditional chemical and physical synthesis of nanoparticles. This study provides a more in-depth debate, and benefits of green synthesis, which has ability for development of green research in the future.

Methods for the synthesis of Nanoparticles

There are various methods used for the synthesis of nanoparticles, which are described below in the subsequent paragraphs.

Chemical method

Metallic precursors, stabilizing agents, and reducing agents (both inorganic and organic) are crucial components of the chemical technique. Reducing agents include sodium citrate, ascorbate, sodium borohydride (NaBH₄), elemental hydrogen, polyol process, tollens reagent, N, N-dimethylformamide (DMF), and poly (ethylene glycol)-block copolymers [26].

Physical method

The physical approach used to create NPs consists of a "top-down" strategy in which the material reduces in size use various physical procedures like ultrasonication, microwave (MW) irradiation, and electrochemical treatment. A tube heater with barometrical weight is utilized in this approach to integrate NPs via evaporation condensation. Evaporation condensation and removal with lasers are the two most important physical processes. Inside a pontoon centered around the heater, the source element is vaporized into a gas called the carrier. Previously, several NPs of Ag, Au, PbS, and Cd were synthesized and reported utilizing this dissipation build-up approach [27].

Green synthesis method

The biological methodology, provided as an alternate to chemical and physical methods, provides an environmentally friendly method of generating nanoparticles. Furthermore, no expensive, harmful, or deadly substances are used in this technique. A biological approach that has lately gained popularity may be used to create nanoparticles of various shapes, sizes, arrangements, and physicochemical properties. By using living organisms such as bacteria, actinobacteria, yeasts, moulds, algae, and plants, synthesis can be accomplished in a single step. Plants and microbes use enzymes, protein, phenolic compounds, amines, alkaloids, and pigments to produce nanoparticles via reduction [28-32].

In traditional chemical and physical methods, reducing reagents and stabilizing agents are required to avoid undesirable clumping of the manufactured nanoparticles, which is detrimental to the environment. Furthermore, the nanoparticles produced by these technologies pose a risk due to their composition, size, and

surface chemistry. Nanoparticles with biocompatibility are manufactured using the green synthesis approach; these compounds are present naturally in living organisms [12].

Green synthesis from bacteria and actinobacteria

Bacteria are ideal candidates for the production of nanoparticles due to their rapid growth, low growing costs, and simplicity of control and adaptation to the surrounding environment. Simultaneously, many bacteria species have been revealed to have unique routes for reducing the hazardous properties of metals or chemicals. Bacteria are recommended for these characteristics since they can produce nanoparticles both in and out of the cell. Metal ions can be reduced and precipitated for nanoparticle formation by employing metabolic pathways and reducing agents present inside a bacterial cell, such as proteins and enzymes [23,33].

Actinobacteria are aerobic, inactive, gram-positive bacteria that create additional metabolic products such as antibiotics. They are resistant to the most toxic metallic substances due to their detoxifying abilities. Metal ions that are toxic are detoxified by intracellular or extracellular reduction or aggregation. As a result, antibacterial, antifungal, anticancer, antioxidant, antibiotic-contamination, and catalytic nanoparticles can be produced [34].

Green synthesis using enzymes

Using readily cultured and fast-breeding eukaryotic yeasts and moulds with minimal biomass design, nanoparticles may be created extracellularly or intracellularly with the aid of enzymes. The parameters such as, incubation conditions and metallic ion solutions are used to determine the shape and size of the nanoparticles. Some moulds are toxic to humans, hence, limiting their use in nanoparticle production [35].

Green synthesis from algae

Algae are eukaryotes aquatic photosites which break down metallic ions into nanoparticles using dyes, carbohydrate, lipid, protein, nucleic acid, and secondary metabolic products. The algal extract, which occurs in an aqueous medium at a certain temperature, is supplemented with metal solutions of the suitable pH and concentration, leading to the development of nanoparticles having antibacterial with no hazardous byproducts, are produced. The size of nanoparticles is regulated by factors such as solution incubation time, ambient temperature, mixture pH, and metal ion concentration. Because of the convenience of availability and utility, algae are also favorable to this synthesis approach. Furthermore, when bacteria and plant extracts are utilized, effective biomolecules in the reaction media are less destroyed by the created nanoparticles [36,37].

Vitamin-based green synthesis

A green mixture combination of Ag and palladium nanospheres, nanowires, and nanorods with vitamin B₂ (as reducing and capping agents) has been presented. In the fabrication of nanowires and nanorods, vitamin B₂ acts as a reducing agent. This is a unique strategy in the field of green nanotechnology, indicating the use of natural molecules with anticancer properties [38]. Chitosan is used as a stabilising agent, whereas ascorbic acid is used as a capping and reducing agent. Because chitosan binds with metal ions, the number of NPs produced is directly proportional to the chitosan quantity

used [39]. Malassis L et al. reported a simple method for manufacturing uniformly sized NPs using ascorbic acid and capping material [40]. Furthermore, water-soluble Antioxidants like ascorbic acid seems to be the reason for the decrease in Ag NPs in *Desmodium triflorum*. During glycolysis, plants produce a large amount of H ions, which act as a potent reducing agent and likely to be effective in the production of Ag NPs [41].

Plants and phytochemicals

Plant-based NP synthesis is really feasible. As a result, they are an affordable and critical choice for large-scale NP manufacturing [42]. A Research on the antioxidant properties of extracts of blackberry, blueberry, turmeric, and pomegranate revealed that pomegranate has the ability to produce more homogenous dimensions and shapes of Au and Ag NPs in the 20-500 nm range. These nanoparticles have the potential to be used in the treatment of cancer and preventative medicine [43]. To reduce the platinum compound, *F. herba* isolate was used; the closeness of hydrogen and carbonyl in polyphenolic compounds predominantly works as a fixing agent for metal particles [44].

A number of factors including the nature of the plant extracts, NP production in salt solution may occur quickly; the main causes being the amount of the extracts, metallic salt, pH, and interaction. Dihydroquercetin, quercetin, and rutin reduced AgNO₃ to AgNPs, resulting in the creation of a strong surface plasmon resonance (SPR) band, signifying a reduction of this component [45]. Kou and Varma proposed a simple, green, and quick (5-minute) technique for generating Ag NPs by MW irradiation with beetroot pulp as a reducing reagent. The resulting material demonstrated excellent photocatalytic activity for the degradation of methyl orange (MO) dye [46].

Advantages of Green synthesis of nanoparticles over chemical and physical method

The advantages of using nanoparticles for drug delivery arise from two key factors. First, due to their small size, nanoparticles may travel through narrow capillaries and be taken up by cells, allowing for excellent medication retention at target locations. Secondly, the utilization of biodegradable materials in the creation of nanoparticles allows for prolonged drug distribution inside the target region over several days or weeks [47,48].

Nanoparticles are important not just for medicinal applications, but they also have the ability to revolutionize an extensive spectrum of electrical items and processes. When it comes to electrical equipment, the industries that benefit from the ongoing advancement of nanotechnology include nano diodes, nano transistors, OLED, plasma displays, quantum computers, and many more. Nanotechnology can potentially benefit the energy industry. The batteries, fuel cells, and solar cells, for instance, may be created in tiny quantities yet with greater efficacy using this method. Manufacturing, which utilizes materials like aerogels, nanotubes, nano particles, and other similar commodities to create their products, is another industry that can benefit from nanotechnology. These materials are often more durable, more robust, and cheaper than those created without the use of nanotechnology. There are some additional benefits to nanoparticles in terms of

manufacturing and delivery of drugs. Nanoparticles improved both therapeutic efficacy and bioavailability. They decreased fed/fasted variation which improved medication stability. There is no biotoxicity of the carrier while conveying the medicine with nanoparticles. Nanoparticles provide no problems in large-scale manufacture or sterilization; however, they must be avoided when using organic solvents [47,48].

Conclusion

Several microorganisms and plant compounds are currently employed to successfully synthesize metal nanoparticles for green synthesis. As a result, green nanoparticle synthesis is the most feasible, simple, and sustainable method of creating nanoparticles since it avoids the consumption of poisonous substances and the formation of dangerous undesired byproducts. In this study, we addressed biological NP synthesis techniques that include living organisms such as algae, fungus, and plants. Nanoparticles are widely used due to their exceptional qualities and have received a lot of attention recently. Although, more research is needed to fully exploit the promise of green synthesis, such as optimizing synthesis conditions, investigating novel biomaterials for nanoparticle synthesis, and broadening the variety of novel green synthesis pathways for these ecologically friendly nanoparticles.

Ethical approval

The undertaken research is not associated with the usage of either human or animal use.

AI Disclosure

No AI-assisted tools were used during the write up.

Conflict of interest

The author declares no conflict of interest.

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